## Amendments to the Claims

This listing of claims will replace all prior versions, and listings, of claims in the application:

## **Listing of Claims:**

## 1-3. (canceled)

- 4. (currently amended) A method for manufacturing a lithium-nickel-cobalt-manganese-containing composite oxide represented by a general formula,  $\underline{\text{Li}_p\text{Ni}_x\text{Mn}_{1-x-y}\text{Co}_y\text{O}_{2-q}\text{F}_q}$  LipNixMn1- $\frac{\text{x-yCo}_y\text{O}_{2-q}\text{F}_q}{\text{y}}$  (where  $0.98 \le p \le 1.07$ ,  $0.3 \le x \le 0.5$ ,  $0.1 \le y \le 0.38$ , and  $0 \le q \le 0.05$ ) according to claim 1, comprising:
- a step for synthesizing coagulated particles of a nickel-cobalt-manganese composite hydroxide wherein primary particles obtained by precipitating the nickel-cobalt-manganese composite hydroxide are coagulated to form secondary particles, by supplying an aqueous solution of a nickel-cobalt-manganese salt, an aqueous solution of an alkali-metal hydroxide and an ammonium-ion donor continuously or intermittently to a reaction system, and making the reaction proceed in the <u>a</u> state wherein the temperature of said reaction system is substantially constant within a range between 30 and 70°C, and pH is maintained at a substantially constant value within a range between 10 and 13;
- a step for synthesizing coagulated particles of a nickel-cobalt-manganese composite oxyhydroxide by making an oxidant act on said coagulated composite hydroxide particles; and
- a step for dry-blending at least said coagulated composite oxyhydroxide particles and a lithium salt, and firing the mixture in an oxygen-containing atmosphere.
- 5. (original) The method for manufacturing a lithium-nickel-cobalt-manganese-containing composite oxide according to claim 4, wherein the lithium salt is lithium carbonate.
- 6. (currently amended) A material for a positive electrode active material for a lithium secondary cell consisting of coagulated particles of a nickel-cobalt-manganese composite oxyhydroxide represented by a general formula,  $Ni_xMn_{1-x-y}Co_yOOH$  NixMn1-x-yCoyOOH (where  $0.3 \le x \le 0.5$ , and  $0.1 \le y \le 0.38$ ), formed by synthesizing coagulated particles of a nickel-cobalt-manganese composite hydroxide wherein primary particles obtained by precipitating the

nickel-cobalt-manganese composite hydroxide are coagulated to form secondary particles, by supplying an aqueous solution of a nickel-cobalt-manganese salt, an aqueous solution of an alkalimetal hydroxide and an ammonium-ion donor continuously or intermittently to a reaction system, and making the reaction proceed in the <u>a</u> state wherein the temperature of said reaction system is substantially constant within a range between 30 and 70°C, and pH is maintained at a substantially constant value within a range between 10 and 13; and making an oxidant act on said coagulated composite hydroxide particles.

- 7. (currently amended) The material for a positive electrode active material for a lithium secondary cell according to claim 6, characterized in that the wherein a specific surface area is 4 to  $30 \text{ m}^2/\text{g}$  m<sup>2</sup>/g.
- 8. (currently amended) The material for a positive electrode active material for a lithium secondary cell according to claim 6, characterized in that the wherein a density of the a compressed powder is 2.0 g/cm<sup>3</sup> g/cm<sup>2</sup> or more.
- 9. (currently amended) The material for a positive electrode active material for a lithium secondary cell according to claim 6, characterized in that the wherein a half-value width of the a diffraction peak when  $2\theta$  is  $19 \pm 1^{\circ}$  in X-ray diffraction using Cu-K $\alpha$  lines is 0.3 to 0.5°.
- 10. (currently amended) A method for manufacturing the material for a positive electrode active material for a lithium secondary cell represented by a general formula,  $Ni_xMn_{1-x-y}Co_yOOH$  NixMn1-x-yCoyOOH (where  $0.3 \le x \le 0.5$ , and  $0.1 \le y \le 0.38$ ), according to claim 6, comprising:

a step for synthesizing <u>the</u> coagulated particles of a <u>the</u> nickel-cobalt-manganese composite hydroxide wherein <u>the</u> primary particles obtained by precipitating the nickel-cobalt-manganese composite hydroxide are coagulated to form <u>the</u> secondary particles, by supplying <u>an the</u> aqueous solution of a <u>the</u> nickel-cobalt-manganese salt, <u>an the</u> aqueous solution of <u>an the</u> alkali-metal hydroxide and <u>an the</u> ammonium-ion donor continuously or intermittently to <u>a the</u> reaction system, and making the reaction proceed in the state wherein the temperature of said reaction system is substantially constant within <u>a the</u> range between 30 and 70°C, and pH is maintained at <u>a the</u>

substantially constant value within a the range between 10 and 13; and

a step for synthesizing coagulated particles of a nickel-cobalt-manganese composite oxyhydroxide by making an the oxidant act on said coagulated composite hydroxide particles.